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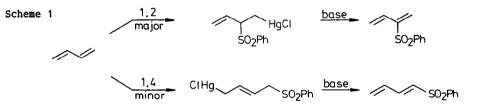
SULFONYLMERCURATION OF CONJUGATED DIENES. A FACILE ROUTE TO ALLYL- AND DIENYL-SULFONES

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<u>Summary</u>: Phenylsulfonylmercuration of 1,3-dienes gives mercury adducts, which on treatment with base afford phenylsulfonyldienes. In most cases the reaction proceeds regioselectively to give 2-(phenylsulfonyl)-1,3-dienes. These are useful synthetic intermediates and can be readily transformed to a variety of functionalized allylsulfones by Michael-type addition.

Addition of nucleophiles and mercury(II) to an alkene (i.e. oxymercuration, solvomercuration etc.) is a well documented methodology in organic synthesis.^{1,2} However, the related additions to conjugated dienes have attracted only a limited interest.³ In connection with our work on metal-catalyzed functionalizations of conjugated dienes^{4,5} we investigated the possibility of a mercury-promoted approach.

Sulfonyl-1,3-dienes are useful synthetic intermediates and recently we⁵ and others⁶ have reported methods for their preparation. In this communication we report a novel route to sulfonyl-1,3-dienes involving phenylsulfonylmercuration^{2b} of 1,3-dienes (Scheme 1).



The phenylsulfonylmercurations were carried out in water at room temperature using a preformed mercury benzenesulfinate complex.⁷ In a typical procedure mercury(II) chloride (13.6 g, 50 mmol) was stirred with sodium benzenesulfinate (12.25 g, 62.5 mmol) in water (160 mL) - Me_2SO (40 mL) at 20 °C for 2 h. 1,3-cyclohexadiene (4.4 g, 55 mmol) was added to the resulting suspension and the mixture was stirred for 2 h at 20 °C, resulting in a white precipitate. The reaction mixture was cooled (ice bath) and the solid material was collected by filtration, washed with water, and dried (desiccator, 0.5 torr) to give 22.1 g (97%) of essentially pure 3-(phenylsulfonyl)-4-(chloromercuri)cyclohexene ($\underline{6}$) as a white powder.

Entry	Diene	Reaction conditions ^b	Mercury adduct	Yield (%) ^C
1	1/1	A, 5 h	HgCl SO2Ph 1	56d
2	1	A, 18 h	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	85
3	\checkmark	B, 5 h	HgCl SO2Ph 3	98d
4	γ	B, 3h	HgCl SO ₂ Ph <u>4</u>	96d
5	\checkmark	A, 24 h	SO_2Ph $HgCi$ SO_2Ph $HgCi$ SO_2Ph $HgCi$ SO_2Ph $HgCi$ SO_2Ph $HgCi$ SO_2Ph $HgCi$	62
6	\bigcirc	B, 2 h	€ ^f SO ₂ Ph	97đ
7	\rightarrow	A, 1 h	SO ₂ Ph Z	82 ^đ
8	\bigcirc	A, 18 h	$\begin{array}{c} & & \\$	56

Table I. Phenylsulfonylmercuration of Conjugated Dienes.^a

a. The reactions were performed at 20 °C in either water or water-Me₂SO (80:20) using NaSO₂Ph and HgCl₂. b. A. Water, B: Water-Me₂SO (80:20). c. Isolated yield. d. >95% 1,2-adduct. e. Only the E-isomer was observed. f. <u>trans</u>-stereochemistry indicated by ¹H NMR g. Mixture of <u>cis</u>- and <u>trans</u>-isomer.

An analytical sample was obtained by recrystallization from ethyl acetate; mp 129 °C (dec.).

Elimination of mercury² to give the sulfonyldiene was accomplished by stirring $\underline{6}$ (4.58 g, 10 mmol) and Na₂CO₃ (4.24 g, 40 mmol) in Me₂SO (80 mL) - water (15 mL) followed by addition of 2 M <u>aqueous NaOH (5 mL)</u>. After stirring at 20 °C for 15 min, 0.5 M aqueous NaHCO₃ (100 mL) was added. Filtration and extraction afforded 1.89 g (86%) of 2-(phenylsulfonyl)-1,3-cyclohexadiene.

The results from phenylsulfonylmercuration of some representative conjugated dienes are given in Table I. In most cases the selectivity for 1,2-addition is high (entries 1, 3, 4, 6 and 7). However, 1,3-cycloheptadiene gave a less regioselective reaction (entry 8), and isoprene and $\underline{E}, \underline{E}-2, 4$ -hexadiene afforded 1,2-/1,4 adducts in a ratio of 87:13 and 83:17 respectively (entries 2 and 5). The high selectivity for 1,2-addition (>95%) to 1,3-pentadiene (entry 3) is contrary to the corresponding oxymercuration of the same diene which has been reported to occur mainly 1,4.^{3a,4} A reasonable explanation for these observations is that sulfonylmercuration occurs under kinetic control, whereas oxymercuration occurs under thermodynamic control.⁴

Elimination of mercury² from the sulfonylmercuration adducts is facile and can be performed with several bases (Table II). The choice of base and reaction conditions depend on the substrate and the stability of the product. For example, transformation of <u>3</u> and <u>4</u> to the sulfonyldienes <u>9</u> and <u>10</u> required great care and had to be carried out in aqueous Me₂SO using NaOH-Na₂CO₃ as base. The elimination of mercury from <u>5</u> and <u>6</u> to yield <u>11</u> and <u>12</u> was less sensitive to the reaction conditions.

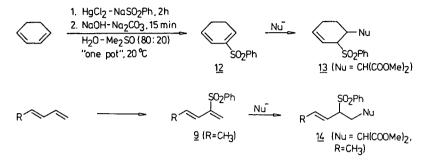
Entry	Mercury compound	Base	Reaction time	Product	yield (%) ^b
1	3	NaOH-Na2CO3c	5 min	SO2Ph <u>9</u>	85
2	4	NaOH-Na ₂ CO ₃ C	15 min	502Ph	70
3	<u>5a</u> + <u>5b</u> (83 : 17)	Na ₂ co ₃ d	4 h	502^{Ph} 502^{Ph} 502^{Ph} 502^{Ph} $11a$ $(83 : 17)$	57
4	<u>6</u>	Na2C03d	4.5 h	\bigtriangledown	89
		NaOH-Na ₂ CO ₃ C Et ₃ N ^e	15 min 24 h	502Ph 12	8 <u>6</u> 76

Table II. Phenylsulfonyl-1,3-dienes from phenylsulfonylmercuration adducts.^a

a. The reactions were performed at 20 °C. b. Isolated yields. c. In Me₂SO-water (80:20).
d. In dioxane-water (50:50). e. In dioxane.

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Since both the addition and elimination steps can be performed in water - Me_2SO , isolation of the mercury adduct is not necessary before the elimination. We have demonstrated this on 1,3-cyclohexadiene which was transformed into sulfonyldiene <u>12</u> in 84% yield in a one-pot procedure.



The 2-(phenylsulfonyl)-1,3-dienes, prepared by the present method, are versatile synthetic intermediates. They can be used in cycloaddition reactions,⁹ and importantly, Michael-type addition of nucleophiles leads to allylic sulfones. A specific example was demonstrated by the addition of dimethyl malonate to <u>12</u> and <u>9</u> to give allylic sulfones <u>13</u> and <u>14</u> in 85 and 75% yield respectively. The allylic sulfones thus obtained are important synthetic intermediates, 6, 10, 11 that allow further functionalization by <u>both</u> electrophiles and nucleophiles. The regioselectivity of these further functionalizations will be investigated.

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